# A NEW STEREOCONTROLLED APPROACH TO A KEY INTERMEDIATE IN THE SYNTHESIS OF (2S,3R)-CAPREOMYCIDINE 

Miroslava Martinkovíal,*, Jozef Gonda ${ }^{a 2}$ and Martina Džoganovíb<br>${ }^{a}$ Department of Organic Chemistry, Institute of Chemistry, P. J. Šafárik University, 04167 Košice, Slovak Republic; e-mail: ${ }^{1}$ mmartin@kosice.upjs.sk, ${ }^{2}$ jgonda@kosice.upjs.sk<br>${ }^{b}$ Institute of Organic Chemistry and Biochemistry, Academy of Sciences of the Czech Republic, 16610 Prague 6, Czech Republic; e-mail: dzoggy@post.sk

Dedicated to Professor Antonín Holy on the occasion of his 70th birthday.

A new stereocontrolled approach to the synthesis of advanced intermediate in the synthesis of nonproteinogenic amino acid ( $2 \mathrm{~S}, 3 \mathrm{R}$ )-capreomycidine via the novel domino reaction has been developed.
Keywords: Capreomycidine; Capreomycins; Stereoselectivity; Nonproteinogenic amino acids; [3,3]-Sigmatropic rearrangements; Total synthesis.

The synthesis of enantiopure nonproteinogenic $\alpha$-amino acids is of great importance due to their role as pharmaceuticals, chiral ligands and building blocks in the synthesis of natural products ${ }^{1}$. Among recent developments in the preparation of such compounds, the functionalization of readily available proteinogenic $\alpha$-amino acids and their derivatives has become an atractive method.

In this paper we demonstrate that D-methionine can be converted very efficiently into pure (2S,3R)-2,3-diacetamido-5-acetoxypentanoic acid (17) as the important intermediate in the stereocontrolled synthesis of (2S,3R)capreomycidine, which is a constituent of the tuberculostatic cyclic peptide antibiotics capreomycins and related tuberactinomycins².

Several syntheses of this nonproteinogenic amino acid in both racemic and also optically pure form have been reported ${ }^{3}$. Recently we published ${ }^{4}$ preliminary communication concerning a new stereocontolled approach to a key precursor for the synthesis of the cyclic guanidino amino acid L-capreomycidine via the novel domino reaction of the [3,3]-sigmatropic rearrangement of chiral thiocyanates easily prepared by chain lengthening

[^0]starting from the corresponding $\alpha$-amino acids followed by stereoselective cyclization ${ }^{5}$.

We now describe a sequence of the reaction steps for the preparation of the protected chiral amino acid 17. As the staring material we have chosen D-methionine which was converted into (R)-2-aminobutano-4-Iactone hydrochloride ${ }^{6}$ (1). Furthermore, tert-butoxycarbonylation of $\mathbf{1}$ with di-tertbutyl dicarbonate in triethylamine, using $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ as the solvent, gave protected lactone 2 ( $92 \%$; Scheme 1), which was then ring-opened using $\mathrm{NaBH}_{4}$ to form N -Boc-D-homoserinol (3) in $92 \%$ yield. Our subsequent strategy consisted in a specific protection of the primary alcoholic group in position 4. We assumed that due to bulky tert-butoxycarbonyl group the two primary alcohol functions of $\mathbf{3}$ could be differentiated easily. But treatment of $\mathbf{3}$ with TBDMSCI/pyridine gave a mixture of three products (two monosilylated regioisomers and one double-protected derivative). A similar situation has been observed employing tert-butyldiphenylsilyl chloride ${ }^{7}$.


## Scheme 1

Therefore we decided for the formation of the oxazolidine ring which was achieved under mild conditions with 2,2-dimethoxypropane (DMP) in acetone at room temperature with $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$ as a catalyst, yielding acetonide 4 in $83 \%$ yield. The benzylation of 4 was carried out with BnBr and NaH in THF to afford 4-O-benzyl derivative 5 (87\%; Scheme 1). Hydrolysis of the acetonide using Amberlite IR120 $\mathrm{H}^{+}$in $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{H}_{2} \mathrm{O}$ gave suitably protected alcohol 6 ( $89 \%$ ) which was subsequently oxidized with
o-iodoxybenzoic acid ${ }^{8}$ (IBX) in DMSO to yield corresponding aldehyde. The aldehyde was used directly in the olefination under Wittig conditions $\left(\mathrm{Ph}_{3} \mathrm{P}=\mathrm{CHCOOCH}_{3}\right)$ giving (E)-ester 7 in $80 \%$ overall yield, the ester was subjected to reduction with diisobutylaluminium hydride to yield allylic alcohol 8 ( $80 \%$ ). The required thiocyanate 9 was easily prepared by the twostep process of mesylation of alcohol $\mathbf{8}$ followed by displacement using KSCN in $\mathrm{CH}_{3} \mathrm{CN}$ in $64 \%$ overall yield. Thermal rearangement of thiocyanate 9 (Scheme 1) proceeded upon heating at $80^{\circ} \mathrm{C}$ in dry xylene in the presence of catalytic amount of 2-hydroxypyridine to afford (4R,5R)-10 as the only isolatable product in $85 \%$ yield. The observed stereochemistry of cyclic thiourea $\mathbf{1 0}$ is in agreement with previous work ${ }^{5}$. The trans product 10 is formed by intramolecular amino addition to syn-isothiocyanate (Scheme 2) which was in the reaction mixture together with anti derivative without selelectivity (syn/anti $\approx 50: 50$ ) as shown by NMR analysis of the crude reactiom mixture after heating at $80^{\circ} \mathrm{C}$ for 35 min when the rearrangement was complete. These diastereoisomers, however, could not be

$$
\begin{aligned}
& \xrightarrow{\text { (i) }}[\text { (ii) } \\
& 15 \\
& \begin{array}{l}
10, \mathrm{R}=\mathrm{Boc} \\
13, \mathrm{R}=\mathrm{H}
\end{array} \square \text { (iv) } \\
& \begin{array}{l}
\text { 12, } R=B o c \square \text { (iv) } \\
\text { 14, } R=H \quad \longleftarrow \text { (in }
\end{array} \\
& 16 \\
& 17 \\
& \text { (i) xylene, } 80^{\circ} \mathrm{C} \text {, 2-hydroxypyridine; (ii) (aminomethyl)cyclopropane, } \mathrm{Et}_{2} \mathrm{O} \text {; (iii) } \mathrm{MNO}, \mathrm{CH}_{3} \mathrm{CN} \text {; } \\
& \text { (iv) } \mathrm{CF}_{3} \mathrm{COOH} / \mathrm{H}_{2} \mathrm{O} 95: 5 \text {; (v) } 6 \mathrm{M} \mathrm{HCl}, \mathrm{~N}_{2} \text {, heating, then pyridine, } \mathrm{Ac}_{2} \mathrm{O} \text {; (vi) } \mathrm{NaIO}_{4}, \mathrm{RuCl}_{3} \text {, } \\
& \mathrm{CCl}_{4} / \mathrm{CH}_{3} \mathrm{CN} / \mathrm{H}_{2} \mathrm{O} \text { 1:1:1; (vii) } \mathrm{NaClO}_{2}, \mathrm{CH}_{3} \mathrm{CN} / t \text { - } \mathrm{BuOH} / 2 \text {-methylbut-2-ene 4:4:1 }
\end{aligned}
$$

[^1]easily separated by chromatography, therefore they were converted into the thioureas 11a and 11b easily differentiable by the reaction with (aminomethyl)cyclopropane (Scheme 2). Our approach to the build-up of chiral amino acid $\mathbf{1 7}$ was based on hydrolysis of cyclic thiourea 10. Treatment of $\mathbf{1 0}$ with 6 m HCl resulted in no reaction. Unprotected cyclic thiourea $\mathbf{1 3}$ was isolated from the reaction mixture as a major product. Removal of the tert-butoxycarbonyl protecting group from $\mathbf{1 0}$ with TFA/ $\mathrm{H}_{2} \mathrm{O}$ provided the same thiourea 13 ( $80 \%$ ). We therefore proceeded with cyclic urea 12, which was prepared from $\mathbf{1 0}$ by the use of mesitylnitrile oxide ${ }^{9}$ (MNO) in acetonitrile in $86 \%$ yield and fully characterized in the form of the unprotected cyclic urea 14 (87\%; Scheme 2). The cyclic urea 12 was converted into the pure suitably protected diamino acid 17 in two reaction steps. First, $\mathbf{1 2}$ was subjected to reaction with 6 m HCl and then the acetylation of dihydrochloride with acetic anhydride in pyridine gave fully protected acetate 15 in $76 \%$ overall yield. Surprisingly, during the hydrolysis ( 6 m HCl , reflux) the benzyl protecting group was removed. The oxidation of $\mathbf{1 5}$ was accomplished with a catalytic amount of ruthenium(III) chloride and $\mathrm{NaIO}_{4}$ in $\mathrm{CCl}_{4} / \mathrm{CH}_{3} \mathrm{CN} / \mathrm{H}_{2} \mathrm{O}$ (1:1:1) to give aldehyde $\mathbf{1 6}$ instead of corresponding acid ${ }^{10}$. Without purification (this product was used immediately in the next step due to instability of $\alpha$-amino aldehydes), the crude $\mathbf{1 6}$ was selectively oxidized to protected diamino acid 17 by treatment with sodium chlorite/ 2-methylbut-2-ene in $73 \%$ yield after flash chromatography (Scheme 2 ).

A stereocontrolled synthesis of chiral non-racemic advanced intermediate 17 in the synthesis of nonproteinogenic amino acid (2S,3R)capreomycidine has been reported emploing D-methionine as a reactant. It was found that the novel domino reaction ${ }^{4,5}$ is a useful strategy for the preparation of diastereomerically pure (4R,5R)-4-vinyItetrahydroimidazole-2-thione 10.

## EXPERIMENTAL

The melting points were determined on the Kofler block and are uncorrected. Optical rotations were measured with a P3002 Krüss polarimeter in chloroform and $\mathrm{H}_{2} \mathrm{O}$ and reported as follows: $[\alpha]_{D}{ }^{25}$ (c in g/100 ml solvent). NMR spectra were recorded at room temperature on a FT NMR spectrometer Varian Mercury Plus $400\left({ }^{1} \mathrm{H}\right.$ at 400.13 MHz and ${ }^{13} \mathrm{C}$ at 100.6 MHz$)$, on a FT NMR spectrometer Bruker AMX $360\left({ }^{1} \mathrm{H}\right.$ at 360.13 MHz and ${ }^{13} \mathrm{C}$ at 90.55), on a FT NMR spectrometer Bruker Avance $500\left({ }^{1} \mathrm{H}\right.$ at 500.13 M Hz and ${ }^{13} \mathrm{C}$ at 125.7 MHz ) and on a FT NMR spectrometer Varian Unity- $500\left({ }^{1} \mathrm{H}\right.$ at 499.8 MHz and ${ }^{13} \mathrm{C}$ at 125.7 MHz$)$. Chemical shifts ( $\delta, \mathrm{ppm}$ ) are referenced either to tetramethylsilane as internal standard for ${ }^{1} \mathrm{H}$ or to the solvent signal ( ${ }^{13} \mathrm{C}$ NMR, $\left.\delta\left(\mathrm{CDCl}_{3}\right) 77.0\right) .{ }^{13} \mathrm{C}$ NMR multiplicities were determined using a DEPT pulse sequence. Coupling constants (J) are given in Hz . IR spectra (wavenumbers in
$\mathrm{cm}^{-1}$ ) were recorded on a Perkin-Elmer 599 IR spectrometer in $\mathrm{CHCl}_{3}$. The reaction course was routinely monitored by TLC (Merck $60 \mathrm{~F}_{254}$ ) and the products were visualized by UV light absorption at 254 nm or by spraying with Mo reagent or $\mathrm{KMnO}_{4}$ reagent. All reactions were performed under an atmosphere of nitrogen when anhydrous solvents were used. Column chromatography was carried out in glass columns using silica gel Kieselgel (0.0350.070 mm ).
(R)-2-Aminobutano-4-Iactone Hydrochloride (1)
${ }^{13} \mathrm{C}$ NMR ( $90 \mathrm{MHz}, \mathrm{D}_{2} \mathrm{O}$ ): $31.0\left(\mathrm{C}-3^{\prime}\right)$; 52.7 ( $\mathrm{C}-2^{\prime}$ ); 71.6 ( $\mathrm{C}-4^{\prime}$ ); 178.7 ( $\mathrm{C}=0$ ). The procedure, m.p. and $[\alpha]_{D}$ were consistent with those reported ${ }^{6} .{ }^{13} \mathrm{C}$ NMR spectroscopic data have not previously been reported ${ }^{6}$.

## (R)-2-[(tert-Butoxycarbonyl)amino]butano-4-Iactone (2)

To a suspension of (R)-2-aminobutano-4-lactone hydrochloride ( $\mathbf{1} ; 4.60 \mathrm{~g}, 33.5 \mathrm{mmol}$ ) in dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}(42 \mathrm{ml})$ were added $\mathrm{Et}_{3} \mathrm{~N}(4.60 \mathrm{ml}, 33.5 \mathrm{mmol})$ and $(\mathrm{BoC})_{2} \mathrm{O}(7.30 \mathrm{~g}, 33.5 \mathrm{mmol})$. After stirring at room temperature for 12 h , the reaction mixture was washed with 1 m aqueous $\mathrm{KHSO}_{4}(40 \mathrm{ml}), 1 \mathrm{~m}$ aqueous $\mathrm{NaHCO}_{3}(40 \mathrm{ml})$, brine ( 20 ml ) and dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$. Evaporation of the solvent at reduced pressure and chromatography of the residue (dichloro-methane-methanol, $95: 5$ ) afforded 6.20 g (92\%) of compound 2 as a white solid, m.p. $115-118{ }^{\circ} \mathrm{C}$ (dichloromethane-methanol), $[\alpha]_{D}{ }^{25}$-13.6 (c $0.23, \mathrm{CHCl}_{3}$ ). (Ref. ${ }^{11}$ gives $[\alpha]_{D}{ }^{21}$ -9.9 (c 1.3, $\mathrm{CHCl}_{3}$ ) but ${ }^{13} \mathrm{C}$ NMR data were not correct in this literature; in ref. ${ }^{12}{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data are published but without $[\alpha]_{D}$ ). For $\mathrm{C}_{9} \mathrm{H}_{15} \mathrm{NO}_{4}$ (201.2) calculated: $53.72 \% \mathrm{C}$, $7.51 \% \mathrm{H}, 6.96 \% \mathrm{~N}$; found: $53.51 \% \mathrm{C}, 7.30 \% \mathrm{H}, 6.72 \%$ N. ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 1.46 \mathrm{~s}$, $9 \mathrm{H}\left(3 \times \mathrm{CH}_{3}\right) ; 2.20 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-3^{\prime}\right) ; 2.76 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-3^{\prime}\right) ; 4.25 \mathrm{ddd}, 1 \mathrm{H}, \mathrm{J}\left(4^{\prime}, 4^{\prime}\right)=9.3$, J(4', $\left.3^{\prime}\right)=$ 5.9, J $\left(4^{\prime}, 3^{\prime}\right)=2.1\left(\mathrm{H}-4^{\prime}\right) ; 4.36 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-2^{\prime}\right) ; 4.45 \mathrm{~m}, 1 \mathrm{H}(\mathrm{H}-4) ; 5.09 \mathrm{~s}, 1 \mathrm{H}(\mathrm{NH}) .{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 28.1\left(3 \times \mathrm{CH}_{3}\right) ; 31.0\left(\mathrm{C}-3^{\prime}\right) ; 49.9\left(\mathrm{C}-2^{\prime}\right) ; 65.5\left(\mathrm{C}-4^{\prime}\right) ; 80.3\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right) ; 155.3$ ( $\mathrm{C}=0$ ); 175.2 ( $\mathrm{C}-1^{\prime}$ ).
(R)-2-[(tert-Butoxycarbonyl)amino]butane-1,4-diol (3)

To a solution of Iactone $2(6.10 \mathrm{~g}, 30.3 \mathrm{mmol})$ in dry THF ( 74 ml ) was added $\mathrm{NaBH}_{4}(2.47 \mathrm{~g}$, $60.6 \mathrm{mmol})$. The reaction mixture was refluxed for 1 h , cooled and quenched with methanol $(3 \mathrm{ml})$. The mixture was poured into saturated aqueous $\mathrm{NaHCO}_{3}(40 \mathrm{ml})$, the product was extracted with ethyl acetate $(3 \times 40 \mathrm{ml})$. The combined organic layers were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated at reduced pressure. The residue was chromatographed on silica gel (cyclo-hexane-ethyl acetate, $1: 2$ ) and afforded $5.75 \mathrm{~g}(92 \%)$ of diol 3 as a white solid, m.p. $64-68{ }^{\circ} \mathrm{C}$, $[\alpha]_{D}{ }^{25}+9.9$ (c $0.95, \mathrm{CHCl}_{3}$ ). (Ref. ${ }^{13}$ gives $[\alpha]_{D}{ }^{25}+30.8$ (c $0.4, \mathrm{MeOH}$ ), Aldrich handbook for (S)-isomer $[\alpha]_{D}{ }^{20}-8$ (c 1, $\mathrm{CHCl}_{3}$ ); refs ${ }^{12,14}$ do not report $[\alpha]_{D}$ ). For $\mathrm{C}_{9} \mathrm{H}_{19} \mathrm{NO}_{4}$ (205.3) calculated: $52.67 \%$ C, $9.33 \% \mathrm{H}, 6.82 \% \mathrm{~N}$; found: $52.81 \% \mathrm{C}, 9.29 \% \mathrm{H}, 6.65 \% \mathrm{~N} .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{D}_{2} \mathrm{O}$ ): $1.45 \mathrm{~s}, 9 \mathrm{H}\left(3 \times \mathrm{CH}_{3}\right) ; 1.61 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-3^{\prime}\right) ; 1.81 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-3^{\prime}\right) ;$ $3.59-3.72 \mathrm{~m}, 4 \mathrm{H}\left(\mathrm{H}-1^{\prime}, \mathrm{H}-4^{\prime}\right) ; 3.80 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-2^{\prime}\right) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $28.4(3 \times$ $\left.\mathrm{CH}_{3}\right) ; 34.8\left(\mathrm{C}-3^{\prime}\right) ; 49.5\left(\mathrm{C}-2^{\prime}\right) ; 58.7\left(\mathrm{C}-4^{\prime}\right) ; 65.1\left(\mathrm{C}-1^{\prime}\right) ; 80.0\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right) ; 157.1(\mathrm{C}=0)$.
tert-Butyl (4R)-4-(2-Hydroxyethyl)-2,2-dimethyl-1,3-oxazolidine-3-carboxylate (4)
Diol 3 ( $5.70 \mathrm{~g}, 27.80 \mathrm{mmol}$ ) was dissolved in a mixture of acetone ( 100 ml ) and 2,2-dimethoxypropane ( 30 ml ) to which $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}(0.20 \mathrm{ml}$ ) was added. The resulting solution was stirred at room temperature for 1 h (TLC showed no starting 3). The solvent was removed under reduced pressure, residual oil taken up in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(100 \mathrm{ml})$ and the resulting solution was washed with a mixture of saturated aqueous $\mathrm{NaHCO}_{3}$ and $\mathrm{H}_{2} \mathrm{O}(60 \mathrm{ml}, 1: 1)$, then brine ( 60 ml ), dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and the solvent was evaporated. Chromatography of the residue (cyclohexane-ethyl acetate, 2:1) gave 5.63 g ( $83 \%$ ) of compound 4 as a white solid, m.p. $74-76{ }^{\circ} \mathrm{C},[\alpha]_{D}{ }^{25}+13.4$ (c 1.8, $\mathrm{CHCl}_{3}$ ). For $\mathrm{C}_{12} \mathrm{H}_{23} \mathrm{NO}_{4}$ (245.3) calculated: $58.75 \% \mathrm{C}$, $9.45 \% \mathrm{H}, 5.71 \% \mathrm{~N}$; found: $58.83 \% \mathrm{C}, 9.34 \% \mathrm{H}, 5.82 \% \mathrm{~N} .{ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): 1.49 s , $12 \mathrm{H}\left(4 \times \mathrm{CH}_{3}\right) ; 1.53 \mathrm{~s}, 3 \mathrm{H}\left(\mathrm{CH}_{3}\right) ; 1.71 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-3^{\prime}\right) ; 1.82 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-3^{\prime}\right) ; 3.53 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-4^{\prime}\right)$; $3.62 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-2^{\prime}\right) ; 3.68 \mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=8.7(\mathrm{OH}) ; 3.99 \mathrm{~m} 2 \mathrm{H}\left(\mathrm{H}-1^{\prime}\right) ; 4.21 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-4^{\prime}\right)$. ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}, 50^{\circ} \mathrm{C}$ ): $24.4\left(\mathrm{CH}_{3}\right) ; 27.8\left(\mathrm{CH}_{3}\right) ; 28.4\left(3 \times \mathrm{CH}_{3}\right) ; 37.7\left(\mathrm{C}-3^{\prime}\right) ; 54.1$ (C-2'); $58.7\left(\mathrm{C}-4^{\prime}\right) ; 68.3\left(\mathrm{C}-1^{\prime}\right) ; 80.9\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right)$; $93.7(\mathrm{OCN}) ; 153.8(\mathrm{C}=0) .{ }^{1} \mathrm{H}$ NMR spectroscopic data were consistent with those reported in refs ${ }^{14 \mathrm{~b}, 15} .{ }^{13} \mathrm{C}$ NMR data have not previously been reported in refs ${ }^{14 \mathrm{~b}, 15}$.
tert-Butyl (4R)-4-[2-(Benzyloxy)ethyl]-2,2-dimethyl-1,3-oxazolidine-3-carboxylate (5)
To a solution of alcohol $4(4.20 \mathrm{~g}, 17.12 \mathrm{mmol})$ in dry THF ( 34 ml ) at $0^{\circ} \mathrm{C}$ was added NaH ( $1.37 \mathrm{~g}, 34.32 \mathrm{mmol}, 60 \%$ dispersion in mineral oil, freed of oil with anhydrous THF). The reaction was stirred at $0{ }^{\circ} \mathrm{C}$ for 20 min and then $\mathrm{BnBr}(3.10 \mathrm{ml}, 25.68 \mathrm{mmol})$ was added at the same temperature. The mixture was allowed to warm to room temperature and stirred for 12 h and then partitioned between $\mathrm{CH}_{2} \mathrm{Cl}_{2}(50 \mathrm{ml}$ ) and ice water ( 50 ml ). The aqueous phase was extracted with another portion of $\mathrm{CH}_{2} \mathrm{Cl}_{2}(50 \mathrm{ml})$. The combined organic phases were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and the solvent evaporated under reduced pressure. The chromatography of residue on silica gel (cyclohexane-ethyl acetate, 9:1) afforded $5.00 \mathrm{~g}(87 \%)$ of compound 5 as a colorless oil, $[\alpha]_{D}{ }^{25}+16$ (c $0.24 \mathrm{CHCl}_{3}$ ). For $\mathrm{C}_{19} \mathrm{H}_{29} \mathrm{NO}_{4}$ (335.5) calculated: 68.03\% C, 8.71\% H, 4.18\% N; found: 68.14\% C, 8.61\% H, 4.27\% N. ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right): 1.47 \mathrm{~s}, 12 \mathrm{H}\left(4 \times \mathrm{CH}_{3}\right) ; 1.57 \mathrm{~s}, 3 \mathrm{H}\left(\mathrm{CH}_{3}\right) ; 1.85 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-3^{\prime}\right) ; 2.06 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-3^{\prime}\right) ;$ $3.53 \mathrm{~s}, 2 \mathrm{H}\left(\mathrm{H}-4^{\prime}\right) ; 3.93 \mathrm{~m}, 3 \mathrm{H}\left(\mathrm{H}-1^{\prime}, \mathrm{H}-2^{\prime}\right) ; 4.47 \mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=12.4\left(\mathrm{CH}_{2} \mathrm{Ph}\right) ; 4.51 \mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=$ $12.4\left(\mathrm{CH}_{2} \mathrm{Ph}\right) ; 7.26-7.34 \mathrm{~m}, 5 \mathrm{H}(\mathrm{Ph}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $24.6\left(\mathrm{CH}_{3}\right) ; 27.6\left(\mathrm{CH}_{3}\right)$; $28.5\left(3 \times \mathrm{CH}_{3}\right) ; 33.8\left(\mathrm{C}-3^{\prime}\right) ; 56.3\left(\mathrm{C}-2^{\prime}\right) ; 67.4\left(\mathrm{C}-1^{\prime}\right) ; 68.0\left(\mathrm{C}-4^{\prime}\right) ; 72.9\left(\mathrm{CH}_{2} \mathrm{Ph}\right) ; 80.0\left(\left(\mathrm{CH}_{3}\right)_{3} \mathbf{C}\right)$; 93.5 (OCN); 127.5 ( $3 \times$ C-arom.); 128.3 ( $2 \times$ C-arom.); 138.4 (C-arom.); 152.2 ( $\mathrm{C}=0$ ).

## tert-Butyl N-[(1R)-3-(Benzyloxy)-1-(hydroxymethyl)propyl]carbamate (6)

To a solution of $5(4.90 \mathrm{~g}, 14.61 \mathrm{mmol})$ in methanol $-\mathrm{H}_{2} \mathrm{O}(9: 1,62 \mathrm{ml})$ was added Amberlite IR-120 $\mathrm{H}^{+}(18.57 \mathrm{~g})$. The mixture was stirred at room temperature overnight, filtered and the solid was washed with methanol. Evaporation of the solvent from the combined filtrates gave an oil which was purified by chromatography on silica gel (cyclohexane-ethyl acetate, 3:1) to afford $3.85 \mathrm{~g}(89 \%)$ of the alcohol 6 as a colorless oil, $[\alpha]_{D}{ }^{25}+15$ (c $0.23, \mathrm{CHCl}_{3}$ ). For $\mathrm{C}_{16} \mathrm{H}_{25} \mathrm{NO}_{4}$ (295.4) calculated: $65.06 \% \mathrm{C}, 8.53 \% \mathrm{H}, 4.74 \% \mathrm{~N}$; found: $64.83 \% \mathrm{C}, 8.59 \% \mathrm{H}$, $4.62 \% \mathrm{~N} .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $1.44 \mathrm{~s}, 9 \mathrm{H}\left(3 \times \mathrm{CH}_{3}\right) ; 1.77-1.94 \mathrm{~m}, 2 \mathrm{H}\left(\mathrm{H}-3^{\prime}\right)$; $3.25 \mathrm{bs}, 1 \mathrm{H}(\mathrm{OH}) ; 3.57-3.62 \mathrm{~m}, 4 \mathrm{H}\left(\mathrm{H}-1^{\prime}, \mathrm{H}-4^{\prime}\right) ; 3.76 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-2^{\prime}\right) ; 4.46 \mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=12.3$ $\left(\mathrm{CH}_{2} \mathrm{Ph}\right) ; 4.49 \mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=12.3\left(\mathrm{CH}_{2} \mathrm{Ph}\right) ; 5.23 \mathrm{bs}, 1 \mathrm{H}(\mathrm{NH}) ; 7.24-7.37 \mathrm{~m}, 5 \mathrm{H}(\mathrm{Ph}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $28.5\left(3 \times \mathrm{CH}_{3}\right) ; 31.5\left(\mathrm{C}-3^{\prime}\right) ; 51.3\left(\mathrm{C}-2^{\prime}\right) ; 65.7\left(\mathrm{C}-4^{\prime}\right.$ or $\left.\mathrm{C}-1^{\prime}\right) ; 67.3$ (C-4' or

C-1'); $73.4\left(\mathrm{CH}_{2} \mathrm{Ph}\right) ; 79.5\left(\left(\mathrm{CH}_{3}\right)_{3} \mathbf{C}\right) ; 127.9(3 \times \mathrm{C}$-arom. $) ; 128.6(2 \times \mathrm{C}$-arom.) 137.8 (C-arom.); 156.4 ( $\mathrm{C}=0$ ).

Methyl (2E,4R)-6-(Benzyloxy)-4-[(tert-butoxycarbonyl)amino]hex-2-enoate (7)
To a solution of $6(3.80 \mathrm{~g}, 12.86 \mathrm{mmol})$ in DM SO ( 29 ml ) was added IBX ( $4.00 \mathrm{~g}, 14.15 \mathrm{mmol}$ ). The reaction mixture was stirred at room temperature for 5 h . Ice water ( 90 ml ) was added and the precipitate was filtered off. The filtrate was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \times 90 \mathrm{ml})$. The combined organic phases were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and the solvent was removed under reduced pressure. To the aldehyde dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(34 \mathrm{ml})$, [(methoxycarbonyl)methylidene]triphenylphosphorane ( $5.16 \mathrm{~g}, 15.43 \mathrm{mmol}$ ) was added. The reaction mixture was stirred at room temperature overnight. The solvent was removed at reduced pressure and the residue was chromatographed on silica gel (cyclohexane-ethyl acetate, 5:1) to give 3.60 g ( $80 \%$ ) of ester 7 as a colorless oil, $[\alpha]_{D}{ }^{25}+53$ (c $0.09, \mathrm{CHCl}_{3}$ ). For $\mathrm{C}_{19} \mathrm{H}_{27} \mathrm{NO}_{5}$ (349.4) calculated: $65.31 \% \mathrm{C}, 7.79 \% \mathrm{H}, 4.01 \% \mathrm{~N}$; found: $65.14 \% \mathrm{C}, 7.63 \% \mathrm{H}, 4.14 \% \mathrm{~N} .{ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right): 1.44 \mathrm{~s}, 9 \mathrm{H}\left(3 \times \mathrm{CH}_{3}\right) ; 1.79 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-5^{\prime}\right) ; 1.99 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-5^{\prime}\right) ; 3.55 \mathrm{~m}, 2 \mathrm{H}\left(\mathrm{H}-6^{\prime}\right) ;$ $3.72 \mathrm{~s}, 3 \mathrm{H}\left(\mathrm{CH}_{3} \mathrm{O}\right) ; 4.48 \mathrm{~m}, 2 \mathrm{H}\left(\mathrm{CH}_{2} \mathrm{Ph}\right) ; 4.49 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-4^{\prime}\right) ; 5.37 \mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=6.6(\mathrm{NH}) ;$ $5.93 \mathrm{dd}, 1 \mathrm{H}, \mathrm{J}\left(3^{\prime}, 2^{\prime}\right)=15.7, \mathrm{~J}\left(4^{\prime}, 2^{\prime}\right)=1.7\left(\mathrm{H}-2^{\prime}\right) ; 6.87 \mathrm{dd}, 1 \mathrm{H}, \mathrm{J}\left(3^{\prime}, 2^{\prime}\right)=15.7$, J $\left(4^{\prime}, 3^{\prime}\right)=4.9$ (H-3'); 7.27-7.37 m, $5 \mathrm{H}(\mathrm{Ph}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $28.4\left(3 \times \mathrm{CH}_{3}\right) ; 33.6\left(\mathrm{C}-5^{\prime}\right) ; 50.2$ $\left.\left(\mathrm{C}-4^{\prime}\right) ; 51.6\left(\mathrm{CH}_{3} \mathrm{O}\right) ; 67.0\left(\mathrm{C}-6^{\prime}\right) ; 73.3\left(\mathrm{CH}_{2} \mathrm{Ph}\right) ; 79.5\left(\left(\mathrm{CH}_{3}\right)\right)_{3} \mathrm{C}\right) ; 120.6\left(\mathrm{C}-2^{\prime}\right) ; 127.7(2 \times$ C-arom.); 127.8 (C-arom.); 128.5 ( $2 \times$ C-arom.); 137.8 (C-arom.); 148.5 (C-3'); 155.2 ( $\mathrm{C}=0$ ); $166.7(C=0)$.
tert-Butyl N-\{(1R,2E)-1-[2-(Benzyloxy)ethyl]-4-hydroxybut-2-en-1-yl \}carbamate (8)
To a solution of ester $7(3.50 \mathrm{~g}, 10.02 \mathrm{mmol})$ in dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}(47 \mathrm{ml})$ was added $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$ $(0.19 \mathrm{ml})$ at $-10^{\circ} \mathrm{C}$. Then diisobutylaluminium hydride ( $30.15 \mathrm{ml}, 1.2 \mathrm{~m}$ toluene solution) was dropped into cooled solution, the resulting mixture was stirred at $-10^{\circ} \mathrm{C}$ for 1 h and then quenched with methanol ( 7.5 ml ). The mixture was allowed to warm to room temperature and poured into $30 \%$ aqueous KNa tartarate ( 157 ml ). After stirring for 30 min , the product was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 70 \mathrm{ml})$. The combined organic layers were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and the solvent evaporated under reduced pressure. Chromatography of the residue (cyclohexane-ethyl acetate, 1:1) afforded 2.58 g ( $80 \%$ ) of allylic alcohol 8 as a colorless oil, $[\alpha]_{D}{ }^{25}+61$ (c $0.18, \mathrm{CHCl}_{3}$ ). For $\mathrm{C}_{18} \mathrm{H}_{27} \mathrm{NO}_{4}$ (321.4) calculated: $67.26 \% \mathrm{C}, 8.47 \% \mathrm{H}$, $4.36 \% \mathrm{~N}$; found: $67.01 \% \mathrm{C}, 8.22 \% \mathrm{H}, 4.48 \% \mathrm{~N} .{ }^{1} \mathrm{H} \mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 1.44 \mathrm{~s}, 9 \mathrm{H}(3 \times$ $\left.\mathrm{CH}_{3}\right) ; 1.74 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-5^{\prime}\right) ; 1.91 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-5^{\prime}\right) ; 3.50-3.61 \mathrm{~m}, 3 \mathrm{H}\left(\mathrm{H}-6^{\prime}, \mathrm{OH}\right) ; 4.07 \mathrm{dd}, 1 \mathrm{H}$, $\mathrm{J}\left(1^{\prime}, 1^{\prime}\right)=12.0, \mathrm{~J}\left(2^{\prime}, 1^{\prime}\right)=5.3\left(\mathrm{H}-1^{\prime}\right) ; 4.09 \mathrm{dd}, 1 \mathrm{H}, \mathrm{J}\left(1^{\prime}, 1^{\prime}\right)=12.0, \mathrm{~J}\left(2^{\prime}, 1^{\prime}\right)=5.0\left(\mathrm{H}-1^{\prime}\right) ; 4.28 \mathrm{~m}$, $1 \mathrm{H}\left(\mathrm{H}-4^{\prime}\right) ; 4.48 \mathrm{~s}, 2 \mathrm{H}\left(\mathrm{CH}_{2} \mathrm{Ph}\right) ; 5.20 \mathrm{bs}, 1 \mathrm{H}(\mathrm{NH}) ; 5.61 \mathrm{dd}, 1 \mathrm{H}, \mathrm{J}\left(3^{\prime}, 2^{\prime}\right)=15.4, \mathrm{~J}\left(4^{\prime}, 3^{\prime}\right)=5.1$ (H-3'); 5.74 ddd, $1 \mathrm{H}, \mathrm{J}\left(3^{\prime}, 2^{\prime}\right)=15.4, \mathrm{~J}\left(2^{\prime}, 1^{\prime}\right)=5.3, \mathrm{~J}\left(2^{\prime}, 1^{\prime}\right)=5.0\left(\mathrm{H}-2^{\prime}\right) ; 7.28-7.37 \mathrm{~m}, 5 \mathrm{H}$ (Ph). ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $28.4\left(3 \times \mathrm{CH}_{3}\right) ; 34.6\left(\mathrm{C}-5^{\prime}\right) ; 50.3\left(\mathrm{C}-4^{\prime}\right) ; 62.9\left(\mathrm{C}-1^{\prime}\right) ; 67.2\left(\mathrm{C}-6^{\prime}\right) ;$ $73.2\left(\mathrm{CH}_{2} \mathrm{Ph}\right) ; 79.2\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right) ; 127.7(3 \times \mathrm{C}$-arom.); $128.4(2 \times \mathrm{C}$-arom.); 129.6 (C-2'); 131.8 ( $\mathrm{C}-3^{\prime}$ ); 138.1 (C-arom.); 155.5 ( $\mathrm{C}=0$ ).
tert-Butyl N -\{(1R,2E)-1-[2-(Benzyloxy)ethyl]-4-thiocyanatobut-2-en-1-yl \}carbamate (9)
To a solution of alcohol $8(2.00 \mathrm{~g}, 6.22 \mathrm{mmol})$ in dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}(21.70 \mathrm{ml})$ were added $\mathrm{Et}_{3} \mathrm{~N}$ $(1.30 \mathrm{ml}, 9.33 \mathrm{mmol})$ and $\mathrm{CH}_{3} \mathrm{SO}_{2} \mathrm{Cl}(0.58 \mathrm{ml}, 7.46 \mathrm{mmol})$ at $0^{\circ} \mathrm{C}$. The reaction mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 15 min and then at room temperature for 2 h . The solvent was evaporated
under reduced pressure. The residue was diluted with diethyl ether ( 40 ml ) and the solid was removed by filtration. The solvent was evaporated to afford the crude mesylate which was used in the subsequent reaction directly without further purification.

The crude mesylate was dissolved in $\mathrm{CH}_{3} \mathrm{CN}(22 \mathrm{ml}) \mathrm{KSCN}(0.90 \mathrm{~g}, 9.33 \mathrm{mmol})$ was added. After stirring at room temperature for 1 h , the solvent was evaporated. The residue was diluted with diethyl ether ( 45 ml ) and the solid was removed by filtration. Evaporation of the solvent at reduced pressure and chromatography of the residue (cyclohexane-ethyl acetate, $5: 1$ ) afforded $1.44 \mathrm{~g}(64 \%)$ of thiocyanate 9 as a colorless oil, $[\alpha]_{D}{ }^{25}+208$ (c 0.15 , $\mathrm{CHCl}_{3}$ ). For $\mathrm{C}_{19} \mathrm{H}_{26} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S}(362.5)$ calculated: $62.96 \% \mathrm{C}, 7.23 \% \mathrm{H}, 7.73 \% \mathrm{~N}, 8.85 \% \mathrm{~S}$; found: $63.06 \% \mathrm{C}, 7.38 \% \mathrm{H}, 7.89 \% \mathrm{~N}, 8.59 \% \mathrm{~S} .{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 1.44 \mathrm{~s}, 9 \mathrm{H}\left(3 \times \mathrm{CH}_{3}\right)$; $1.79 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-5^{\prime}\right) ; 1.95 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-5^{\prime}\right) ; 3.47-3.67 \mathrm{~m}, 4 \mathrm{H}\left(\mathrm{H}-1^{\prime}, \mathrm{H}-6^{\prime}\right) ; 4.35 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-4^{\prime}\right) ;$ $4.47 \mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.8\left(\mathrm{CH}_{2} \mathrm{Ph}\right) ; 4.51 \mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.8\left(\mathrm{CH}_{2} \mathrm{Ph}\right) ; 5.35 \mathrm{bs}, 1 \mathrm{H}(\mathrm{NH}) ; 5.64-5.76 \mathrm{~m}$, $2 \mathrm{H}\left(\mathrm{H}-2^{\prime}, \mathrm{H}-3^{\prime}\right) ; 7.26-7.34 \mathrm{~m}, 5 \mathrm{H}(\mathrm{Ph}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 28.3\left(3 \times \mathrm{CH}_{3}\right) ; 34.0$ (C-5'); 35.7 (C-1'); $50.2\left(\mathrm{C}-4^{\prime}\right) ; 66.9\left(\mathrm{C}-6^{\prime}\right) ; 73.1\left(\mathrm{CH}_{2} \mathrm{Ph}\right) ; 79.2\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right) ; 111.8(\mathrm{SCN}) ; 122.7$ (C-2'); 127.6 ( $2 \times$ C-arom.); 127.7 (C-arom.); 128.3 ( $2 \times \mathrm{C}$-arom.); 138.0 (C-3'); 138.1 (C-arom.); 155.2 ( $\mathrm{C}=0$ ).
tert-Butyl (4R,5R)-4-[2-(Benzyloxy)ethyl]-2-thioxo-5-vinylimidazolidine-1-carboxylate (10)
To a solution of thiocyanate 9 ( $1.35 \mathrm{~g}, 3.72 \mathrm{mmol}$ ) in dry xylene ( 32 ml ) was added 2-hydroxypyridine ( $35.4 \mathrm{mg}, 0.372 \mathrm{mmol}$ ). The reaction mixture was heated at $80^{\circ} \mathrm{C}$ for 5 h under nitrogen atmosphere. The solvent was evaporated at reduced pressure, the chromatography of residue on silica gel (cyclohexane-ethyl acetate, 5:1) gave 1.15 g ( $85 \%$ ) of imidazolidine 10 as a colorless oil, $[\alpha]_{D}{ }^{25}+52$ (c $0.26, \mathrm{CHCl}_{3}$ ). For $\mathrm{C}_{19} \mathrm{H}_{26} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S}$ (362.5) calculated: $62.96 \% \mathrm{C}, 7.23 \% \mathrm{H}, 7.73 \% \mathrm{~N}, 8.85 \% \mathrm{~S}$; found: $63.06 \% \mathrm{C}, \mathrm{H}, 7.38 \% \mathrm{H}, 7.89 \% \mathrm{~N}$, $8.59 \%$ S. IR $\left(\mathrm{CHCl}_{3}\right)$ : $1470(\mathrm{C}=\mathrm{S}) ; 1740(\mathrm{C}=0) ; 3450(\mathrm{NH}) .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $1.52 \mathrm{~s}, 9 \mathrm{H}\left(3 \times \mathrm{CH}_{3}\right) ; 1.98-2.07 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-8^{\prime}\right) ; 2.09-2.16 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-8^{\prime}\right) ; 3.55 \mathrm{ddd}, 1 \mathrm{H}$, $\mathrm{J}\left(9^{\prime}, 9^{\prime}\right)=10.6, \mathrm{~J}\left(9^{\prime}, 8^{\prime}\right)=6.0, \mathrm{~J}\left(9^{\prime}, 8^{\prime}\right)=4.6\left(\mathrm{H}-9^{\prime}\right) ; 3.60 \mathrm{ddd}, 1 \mathrm{H}, \mathrm{J}\left(9^{\prime}, 9^{\prime}\right)=10.6, \mathrm{~J}\left(9^{\prime}, 8^{\prime}\right)=5.6$, $\mathrm{J}\left(9^{\prime}, 8^{\prime}\right)=4.1\left(\mathrm{H}-9^{\prime}\right) ; 4.10 \mathrm{dd}, 1 \mathrm{H}, \mathrm{J}\left(6^{\prime}, 5^{\prime}\right)=6.6, \mathrm{~J}\left(5^{\prime}, 4^{\prime}\right)=5.4\left(\mathrm{H}-5^{\prime}\right) ; 4.25 \mathrm{ddd}, 1 \mathrm{H}, \mathrm{J}\left(8^{\prime}, 4^{\prime}\right)=$ 9.0, J $\left(5^{\prime}, 4^{\prime}\right)=5.4, \mathrm{~J}\left(8^{\prime}, 4^{\prime}\right)=2.7\left(\mathrm{H}-4^{\prime}\right) ; 4.46 \mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.8\left(\mathrm{CH}_{2} \mathrm{Ph}\right) ; 4.50 \mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.8$ $\left(\mathrm{CH}_{2} \mathrm{Ph}\right) ; 5.17 \mathrm{dd}, 1 \mathrm{H}, \mathrm{J}\left(7^{\prime} \mathrm{cis}, 6^{\prime}\right)=10.2$, J(7'cis,7'trans) $=0.7$ (H-7'cis); $5.21 \mathrm{dd}, 1 \mathrm{H}$, $\mathrm{J}\left(7^{\prime}\right.$ trans, $\left.6^{\prime}\right)=17.0, \mathrm{~J}\left(7^{\prime} \mathrm{cis}, 7^{\prime}\right.$ trans $)=0.7$ (H-7'trans); $5.77 \mathrm{ddd}, 1 \mathrm{H}, \mathrm{J}\left(7^{\prime}\right.$ trans, $\left.6^{\prime}\right)=17.0$, $\mathrm{J}\left(7^{\prime} \mathrm{cis}, 6^{\prime}\right)=10.2, \mathrm{~J}\left(6^{\prime}, 5^{\prime}\right)=6.6\left(\mathrm{H}-6^{\prime}\right) ; 7.27-7.36 \mathrm{~m}, 5 \mathrm{H}(\mathrm{Ph}) ; 7.56 \mathrm{~s}, 1 \mathrm{H}(\mathrm{NH}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $28.1\left(3 \times \mathrm{CH}_{3}\right)$; $33.2\left(\mathrm{C}-8^{\prime}\right) ; 60.5\left(\mathrm{C}-5^{\prime}\right) ; 64.5\left(\mathrm{C}-4^{\prime}\right) ; 66.5\left(\mathrm{C}-9^{\prime}\right) ; 73.4$ $\left(\mathrm{CH}_{2} \mathrm{Ph}\right) ; 83.6\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right) ; 117.6\left(\mathrm{C}-7^{\prime}\right) ; 127.8(3 \times \mathrm{C}$-arom.); 128.5 ( $2 \times \mathrm{C}$-arom.); 135.6 (C-6'); 137.9 (C-arom.); 150.0 ( $\mathrm{C}=0$ ); 180.2 ( $\mathrm{C}=\mathrm{S}$ ).

> rel tert-Butyl \{(3R,4S)-1-(Benzyloxy)-4-[3-(cyclopropylmethyl)thioureido]hex-5-en-3-yl\} carbamate (11a) and tert-Butyl \{(3R,4R)-1-(Benzyloxy)-4-[3-(cyclopropylmethyl)thioureido]hex-5-en-3-yl \}carbamate (11b)

A solution of thiocyanate $9(0.25 \mathrm{~g}, 0.69 \mathrm{mmol})$ in dry xylene ( 6 ml ) was heated at $80^{\circ} \mathrm{C}$ for 35 min under nitrogen atmosphere. The solvent was evaporated, the crude residue was dissolved in diethyl ether ( 5 ml ) and (aminomethyl)cyclopropane ( $0.07 \mathrm{ml}, 0.76 \mathrm{mmol}$ ) was added. The reaction mixture was stirred at room temperature for 1.5 h , then concentrated under reduced pressure. Purification of the residue by flash chromatography (cyclohexaneethyl acetate, $7: 1$ ) afforded thioureas 11a ( $0.10 \mathrm{~g}, 33.4 \%$ ) and 11b ( $0.15 \mathrm{~g}, 50.1 \%$ ).

11b: Colorless oil, $[\alpha]_{D}{ }^{25}+197$ (c $0.13, \mathrm{CHCl}_{3}$ ). For $\mathrm{C}_{23} \mathrm{H}_{35} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}$ (433.6) calculated: 63.71\% C, $8.14 \% \mathrm{H}, 9.69 \% \mathrm{~N}, 7.39 \% \mathrm{~S} ;$ found: $63.52 \% \mathrm{C}, ~ 8.21 \% \mathrm{H}, 9.48 \% \mathrm{~N}, 7.54 \% \mathrm{~S}$. ${ }^{1} \mathrm{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 0.25 \mathrm{~m}, 2 \mathrm{H}\left(\mathrm{CH}_{2 \text { cycloprop }}\right) ; 0.54 \mathrm{~m}, 2 \mathrm{H}\left(\mathrm{CH}_{2 \text { cycloprop }}\right) ; 1.06 \mathrm{~m}$, $1 \mathrm{H}\left(\mathrm{CH}_{\text {cycloprop }}\right) ; 1.44 \mathrm{~s}, 9 \mathrm{H}\left(3 \times \mathrm{CH}_{3}\right) ; 1.701 \mathrm{H}, \mathrm{m}\left(\mathrm{H}-2^{\prime}\right) ; 1.98 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-2^{\prime}\right) ; 3.30 \mathrm{~m}, 2 \mathrm{H}$ $\left(\mathrm{CH}_{2} \mathrm{~N}\right) ; 3.57 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-\mathrm{I}^{\prime}\right) ; 3.60 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-4^{\prime}\right) ; 3.62 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-\mathrm{I}^{\prime}\right) ; 3.76 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-3^{\prime}\right)$; $4.49 \mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.7\left(\mathrm{PhCH}_{2} \mathrm{O}\right) ; 4.52 \mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.7\left(\mathrm{PhCH}_{2} \mathrm{O}\right) ; 5.27 \mathrm{bd}, 1 \mathrm{H} \mathrm{J}\left(6^{\prime} \mathrm{cis}, 5^{\prime}\right)=$ 10.6 (H-6'cis); $5.30 \mathrm{bd}, 1 \mathrm{H}, \mathrm{J}\left(6^{\prime}\right.$ trans, $\left.5^{\prime}\right)=17.0$ (H-6'trans); $5.74 \mathrm{ddd}, 1 \mathrm{H}, \mathrm{J}\left(6^{\prime}\right.$ trans, $\left.5^{\prime}\right)=17.0$, $\mathrm{J}\left(6^{\prime}\right.$ cis, $\left.5^{\prime}\right)=10.6, \mathrm{~J}\left(5^{\prime}, 4^{\prime}\right)=6.3\left(\mathrm{H}-5^{\prime}\right) .{ }^{13} \mathrm{C}$ NMR $\left(125.7 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 3.6\left(2 \times \mathrm{CH}_{2 \text { cycloprop }}\right)$; $10.0\left(\mathrm{CH}_{\text {cycloprop }}\right) ; 28.3\left(3 \times \mathrm{CH}_{3}\right) ; 29.7\left(\mathrm{C}-2^{\prime}\right) ; 31.8\left(\mathrm{CH}_{2} \mathrm{NH}\right) ; 52.9\left(\mathrm{C}-3^{\prime}\right) ; 67.7\left(\mathrm{C}-1^{\prime}\right) ; 73.4$ $\left(\mathrm{PhCH}_{2}\right) ; 80.1\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right) ; 118.0\left(\mathrm{C}-6^{\prime}\right) ; 127.7\left(3 \times \mathrm{C}\right.$-arom.); $128.4\left(2 \times \mathrm{C}\right.$-arom); $128.5\left(\mathrm{C}-5^{\prime}\right)$; 137.8 (C-arom.); 157.3 ( $\mathrm{C}=0$ ); 180.9 ( $\mathrm{C}=\mathrm{S}$ ).

11a: Colorless oil, $[\alpha]_{D}{ }^{25}+28$ (c $0.18, \mathrm{CHCl}_{3}$ ). For $\mathrm{C}_{23} \mathrm{H}_{35} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}$ (433.6) calculated: 63.71\% C, 8.14\% H, 9.69\% N, 7.39\% S; found: 63.56\% C, 8.23\% H, 9.52\% N, 7.44\% S. ${ }^{1} \mathrm{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 0.22 \mathrm{~m}, 2 \mathrm{H}\left(\mathrm{CH}_{2 \text { cycloprop }}\right) ; 0.54 \mathrm{~m}, 2 \mathrm{H}\left(\mathrm{CH}_{2 \text { cycloprop }}\right) ; 1.02 \mathrm{~m}$, $1 \mathrm{H}\left(\mathrm{CH}_{\text {cycloprop }}\right) ; 1.44 \mathrm{~s}, 9 \mathrm{H}\left(3 \times \mathrm{CH}_{3}\right) ; 1.78 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-2^{\prime}\right) ; 1.92 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-2^{\prime}\right) ; 3.18 \mathrm{~m}, 2 \mathrm{H}$ $\left(\mathrm{CH}_{2} \mathrm{~N}\right) ; 3.57 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-1^{\prime}\right) ; 3.62 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-1^{\prime}\right) ; 3.74 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-4^{\prime}\right) ; 3.93 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-3^{\prime}\right)$; $4.48 \mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.6\left(\mathrm{PhCH}_{2} \mathrm{O}\right) ; 4.51 \mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=11.6\left(\mathrm{PhCH}_{2} \mathrm{O}\right) ; 5.28 \mathrm{bd}, 1 \mathrm{H}, \mathrm{J}\left(6^{\prime}\right.$ trans, $\left.5^{\prime}\right)=$ 16.8 (H6'trans); $5.31 \mathrm{bd}, 1 \mathrm{H}, \mathrm{J}\left(6^{\prime} \mathrm{cis}, 5^{\prime}\right)=10.5$ (H6'cis); 5.75 ddd, $1 \mathrm{H}, \mathrm{J}\left(6^{\prime}\right.$ trans, $\left.5^{\prime}\right)=16.8$, $\mathrm{J}\left(6^{\prime} \mathrm{cis}, 5^{\prime}\right)=10.5, \mathrm{~J}\left(5^{\prime}, 4^{\prime}\right)=5.9\left(\mathrm{H}-5^{\prime}\right) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(125.7 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 3.6\left(2 \times \mathrm{CH}_{2 \text { cycloprop }}\right)$; $10.0\left(\mathrm{CH}_{\text {cycloprop }}\right) ; 28.3\left(3 \times \mathrm{CH}_{3}\right) ; 29.7\left(\mathrm{C}-2^{\prime}\right) ; 32.0\left(\mathrm{CH}_{2} \mathrm{NH}\right) ; 53.8\left(\mathrm{C}-3^{\prime}\right) ; 67.5\left(\mathrm{C}-1^{\prime}\right) ; 73.4$ $\left(\mathrm{CH}_{2} \mathrm{Ph}\right) ; 80.1\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right) ; 118.0\left(\mathrm{C}-6^{\prime}\right) ; 127.8(3 \times \mathrm{C}$-arom); $128.5(2 \times \mathrm{C}$-arom.) ; 133.4 (C-5'); 137.8 (C-arom.); 157.1 ( $\mathrm{C}=0$ ); 180.9 ( $\mathrm{C}=\mathrm{S}$ ).
tert-Butyl (4R,5R)-4-[2-(Benzyloxy)ethyl]-2-oxo-5-vinylimidazolidine-1-carboxylate (12)
To a solution of $\mathbf{1 0}(0.67 \mathrm{~g}, 1.85 \mathrm{mmol})$ in $\mathrm{CH}_{3} \mathrm{CN}$ ( 19 ml ) was added mesitylnitrile oxide (MNO) ( $0.158 \mathrm{~g}, 0.98 \mathrm{mmol}$ ). The reaction mixture was stirred at room temperature for 15 min , $\mathrm{CH}_{3} \mathrm{CN}$ was evaporated under reduced pressure. The chromatography of residue on silica gel (cyclohexane-ethyl acetate, 3:1) gave 0.55 g ( $86 \%$ ) of 12 as a colorless oil, $[\alpha]_{D}{ }^{25}+105$ (c $0.16, \mathrm{CHCl}_{3}$ ). For $\mathrm{C}_{19} \mathrm{H}_{26} \mathrm{~N}_{2} \mathrm{O}_{4}$ (346.4) calculated: $65.87 \% \mathrm{C}, 7.57 \% \mathrm{H}, 8.09 \% \mathrm{~N}$; found: $65.98 \% \mathrm{C}, 7.36 \% \mathrm{H}, 7.94 \% \mathrm{~N} .{ }^{13} \mathrm{C}$ NMR ( $90 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $28.0\left(3 \times \mathrm{CH}_{3}\right) ; 33.3\left(\mathrm{C}-8^{\prime}\right) ; 55.9$ $\left(\mathrm{C}-5^{\prime}\right) ; 59.6\left(\mathrm{C}-4^{\prime}\right) ; 66.4\left(\mathrm{C}-9^{\prime}\right) ; 73.1\left(\mathrm{CH}_{2} \mathrm{Ph}\right) ; 82.4\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right) ; 116.3\left(\mathrm{C}-7^{\prime}\right) ; 127.5(2 \times$ C-arom.); 127.6 (C-arom.); 128.3 ( $2 \times$ C-arom.); 137.0 ( $\mathrm{C}-6^{\prime}$ ); 137.9 (C-arom.); 150.1 ( $\mathrm{C}=0$ ); $155.6(C=0)$.

## (4R,5R)-4-[2-(Benzyloxy)ethyl]-5-vinylimidazolidine-2-thione (13)

Imidazolidine 10 ( $0.15 \mathrm{~g}, 0.414 \mathrm{mmol}$ ) was dissolved in a mixture of $\mathrm{CF}_{3} \mathrm{COOH}$ and $\mathrm{H}_{2} \mathrm{O}$ ( $95: 5,5 \mathrm{ml}$ ). The reaction mixture was stirred at room temperature for 30 min and then concentrated under reduced pressure. The residue was diluted with diethyl ether ( 10 ml ) and washed with saturated aqueous $\mathrm{NaHCO}_{3}(5 \mathrm{ml})$. The organic phase was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, the solvent was evaporated and the resulting crude product was purified by flash chromatography on silica gel (cyclohexane-ethyl acetate, 1:1) to provide $0.087 \mathrm{~g}(80 \%)$ of 13 as a white solid, m.p. $69-70{ }^{\circ} \mathrm{C},[\alpha]_{D}{ }^{25}+51$ (c $0.86, \mathrm{CHCl}_{3}$ ). For $\mathrm{C}_{14} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{OS}$ (262.4) calculated: 64.09\% C, $6.92 \%$ H, $10.68 \%$ N, $12.22 \%$ S; found: $64.28 \%$ C, $6.76 \%$ H, $10.74 \%$ N, $12.01 \%$ S. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $1.81 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-8^{\prime}\right) ; 1.95 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-8^{\prime}\right) ; 3.52 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-9^{\prime}\right)$; $3.58 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-9^{\prime}\right) ; 3.78 \mathrm{ddd} 1 \mathrm{H}, \mathrm{J}\left(5^{\prime}, 4^{\prime}\right)=8.1, \mathrm{~J}\left(8^{\prime}, 4^{\prime}\right)=5.1$, $\mathrm{J}\left(8^{\prime}, 4^{\prime}\right)=3.6\left(\mathrm{H}-4^{\prime}\right) ; 4.08 \mathrm{dd}$,
$1 \mathrm{H}, \mathrm{J}\left(5^{\prime}, 4^{\prime}\right)=8.1, \mathrm{~J}\left(6^{\prime}, 5^{\prime}\right)=7.7\left(\mathrm{H}-5^{\prime}\right) ; 4.49 \mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=12.2\left(\mathrm{CH}_{2} \mathrm{Ph}\right) ; 4.52 \mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=12.2$ $\left(\mathrm{CH}_{2} \mathrm{Ph}\right) ; 5.23 \mathrm{~d}, 1 \mathrm{H}, \mathrm{J}\left(7^{\prime} \mathrm{cis}, 6^{\prime}\right)=10.1$ (H-7'cis); $5.28 \mathrm{~d}, 1 \mathrm{H}, \mathrm{J}\left(7^{\prime}\right.$ trans, $\left.6^{\prime}\right)=17.5$ (H-7'trans); 5.81 ddd, $1 \mathrm{H}, \mathrm{J}\left(7^{\prime}\right.$ trans, $\left.6^{\prime}\right)=17.5$, J $\left(7^{\prime} \mathrm{cis}, 6^{\prime}\right)=10.1, \mathrm{~J}\left(6^{\prime}, 5^{\prime}\right)=7.7\left(\mathrm{H}-6^{\prime}\right) ; 6.01 \mathrm{bs}, 1 \mathrm{H}(\mathrm{NH})$; $6.50 \mathrm{bs}, 1 \mathrm{H}(\mathrm{NH}) ; 7.29-7.39 \mathrm{~m}, 5 \mathrm{H}(\mathrm{Ph}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): 33.7 (C-8); 62.9 (C-5'); 66.4 (C-4'); 67.5 (C-9'); 73.4 ( $\mathrm{CH}_{2} \mathrm{Ph}$ ); 119.1 (C-7'); 127.8 (2 $\times$ C-arom.); 128.0 (C-arom.); 128.6 (2 $\times$ C-arom.); 135.2, 137.6 (C-arom.); 183.2 ( $\mathrm{C}=\mathrm{S}$ ).
(4R,5R)-4-[2-(Benzyloxy)ethyl]-5-vinylimidazolidin-2-one (14)
Compound 12 ( $0.10 \mathrm{~g}, 0.289 \mathrm{mmol}$ ) was dissolved in a mixture of $\mathrm{CF}_{3} \mathrm{COOH}$ and $\mathrm{H}_{2} \mathrm{O}$ (95:5, 3.50 ml ). The reaction mixture was stirred at room temperature for 20 min and then concentrated under reduced pressure. The residue was diluted with diethyl ether ( 7 ml ) and washed with saturated aqueous $\mathrm{NaHCO}_{3}(3.50 \mathrm{ml})$. The organic phase was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, the solvent was evaporated and the resulting crude product purified by flash chromatography on silica gel (cyclohexane-ethyl acetate, 1:2) to provide $0.062 \mathrm{~g}(87 \%)$ of 13 as a white solid, m.p. $76-78{ }^{\circ} \mathrm{C},[\alpha]_{\mathrm{D}}{ }^{25}+29.4$ (c $0.15, \mathrm{CHCl}_{3}$ ). For $\mathrm{C}_{14} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}_{2}$ (246.3) calculated: $68.27 \% \mathrm{C}, 7.37 \% \mathrm{H}, 11.37 \% \mathrm{~N}$; found: $68.43 \% \mathrm{C}, 7.56 \% \mathrm{H}, 11.14 \% \mathrm{~N} .{ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right): 1.85 \mathrm{~m}, 2 \mathrm{H}\left(\mathrm{H}-8^{\prime}\right) ; 3.56 \mathrm{~m}, 3 \mathrm{H}\left(\mathrm{H}-4^{\prime}, \mathrm{H}-9^{\prime}\right) ; 3.85 \mathrm{dd}, 1 \mathrm{H}, \mathrm{J}\left(5^{\prime}, 4^{\prime}\right)=7.3, \mathrm{~J}\left(6^{\prime}, 5^{\prime}\right)=7.3$ $\left(\mathrm{H}-5^{\prime}\right) ; 4.47 \mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=12.7\left(\mathrm{CH}_{2} \mathrm{Ph}\right) ; 4.50 \mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=12.7\left(\mathrm{CH}_{2} \mathrm{Ph}\right) ; 5.09 \mathrm{bs}, 1 \mathrm{H}(\mathrm{NH})$; 5.16 dd, $1 \mathrm{H}, \mathrm{J}\left(7^{\prime} \mathrm{cis}, 6^{\prime}\right)=10.1$, J(7'cis,7'trans) $=0.9$ (H-7'cis); $5.24 \mathrm{dd}, 1 \mathrm{H}, \mathrm{J}\left(7^{\prime}\right.$ trans, $\left.6^{\prime}\right)=17.1$, $\mathrm{J}\left(7^{\prime} \mathrm{cis}, 7^{\prime}\right.$ trans $)=0.9$ (H-7'trans); $5.25 \mathrm{bs}, 1 \mathrm{H}(\mathrm{NH}) ; 5.82 \mathrm{ddd}, 1 \mathrm{H}, \mathrm{J}\left(7^{\prime}\right.$ trans, $\left.\mathrm{G}^{\prime}\right)=17.1$, $\mathrm{J}\left(7^{\prime} \mathrm{cis}, 6^{\prime}\right)=10.1, \mathrm{~J}\left(6^{\prime}, 5^{\prime}\right)=7.3\left(\mathrm{H}-6^{\prime}\right) ; 7.24-7.36 \mathrm{~m}, 5 \mathrm{H}(\mathrm{Ph}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): 34.3 (C-8'); 57.9 (C-5'); 62.1 (C-4'); 67.8 (C-9'); $73.3\left(\mathrm{CH}_{2} \mathrm{Ph}\right) ; 117.6$ (C-7'); 127.7 ( $2 \times$ C-arom.); 127.8 (C-arom.); 128.5 ( $2 \times \mathrm{C}$-arom.); 137.0 (C-6'); 137.9 (C-arom.); 162.6 ( $\mathrm{C}=0$ ).
(3R,4R)-3,4-Diacetamidohex-5-en-1-yl Acetate (15)
Compound 12 ( $0.41 \mathrm{~g}, 1.183 \mathrm{mmol}$ ) was dissolved in $6 \mathrm{~m} \mathrm{HCl}(20 \mathrm{ml})$ and the solution was heated under reflux for 6 h in nitrogen atmosphere. The solvent was removed under reduced pressure, the resulting solid was diluted with pyridine ( 2 ml ) and $\mathrm{Ac}_{2} \mathrm{O}(0.9 \mathrm{ml}, 9.54$ mmol ) was added. The reaction mixture was stirred at room temperature overnight, the solvent was evaporated and the residue chromatographed on silica gel (dichloromethanemethanol, 95:5) to give $0.23 \mathrm{~g}(76 \%)$ of 15 as a white solid, m.p. $101-102{ }^{\circ} \mathrm{C},[\alpha]_{D}{ }^{25}+30.3$ (c $0.165, \mathrm{CHCl}_{3}$ ). For $\mathrm{C}_{12} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}_{4}$ (256.3) calculated: $56.24 \% \mathrm{C}, 7.87 \% \mathrm{H}, 10.93 \% \mathrm{~N}$; found: $56.40 \% \mathrm{C}, 7.56 \% \mathrm{H}, 11.04 \% \mathrm{~N} .{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 1.98 \mathrm{~s}, 3 \mathrm{H}\left(\mathrm{CH}_{3}\right) ; 1.99 \mathrm{~s}, 3 \mathrm{H}$ $\left(\mathrm{CH}_{3}\right) ; 2.05 \mathrm{~s}, 3 \mathrm{H}\left(\mathrm{CH}_{3}\right) ; 1.74 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-2^{\prime}\right) ; 2.02 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-2^{\prime}\right) ; 4.03 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-3^{\prime}\right) ; 4.13 \mathrm{~m}$, $2 \mathrm{H}\left(\mathrm{H}-1^{\prime}\right) ; 4.41 \mathrm{ddd}, 1 \mathrm{H}, \mathrm{J}\left(4^{\prime}, \mathrm{NH}\right)=8.4, \mathrm{~J}\left(4^{\prime}, 3^{\prime}\right)=7.6, \mathrm{~J}\left(5^{\prime}, 4^{\prime}\right)=7.0\left(\mathrm{H}-4^{\prime}\right) ; 5.24 \mathrm{dd}, 1 \mathrm{H}$, $\mathrm{J}\left(6^{\prime} \mathrm{cis}, 5^{\prime}\right)=10.1$, J(6'cis, $6^{\prime}$ trans) $=0.5$ (H-6'cis); $5.29 \mathrm{dd}, 1 \mathrm{H}, \mathrm{J}\left(6^{\prime}\right.$ trans, $\left.5^{\prime}\right)=17.1$, J( $6^{\prime}$ cis, $6^{\prime}$ trans $)=$ 0.5 (H-6'trans); 5.76 ddd, $1 \mathrm{H}, \mathrm{J}\left(6^{\prime}\right.$ trans, $\left.5^{\prime}\right)=17.1, \mathrm{~J}\left(6^{\prime} \mathrm{cis}, 5^{\prime}\right)=10.1, \mathrm{~J}\left(5^{\prime}, 4^{\prime}\right)=7.0\left(\mathrm{H}-5^{\prime}\right)$; $6.77 \mathrm{~d}, 1 \mathrm{H}, \mathrm{J}\left(3^{\prime}, \mathrm{NH}\right)=8.9(\mathrm{NH}) ; 6.93 \mathrm{~d}, 1 \mathrm{H}, \mathrm{J}\left(4^{\prime}, \mathrm{NH}\right)=8.4(\mathrm{NH}) .{ }^{13} \mathrm{C} \mathrm{NMR}(100 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right): 20.9\left(\mathrm{CH}_{3}\right) ; 23.0\left(\mathrm{CH}_{3}\right) ; 23.1\left(\mathrm{CH}_{3}\right) ; 30.5\left(\mathrm{C}-2^{\prime}\right) ; 50.2\left(\mathrm{C}-3^{\prime}\right) ; 56.4\left(\mathrm{C}-4^{\prime}\right) ; 61.2\left(\mathrm{C}-1^{\prime}\right) ;$ 118.1 ( $\mathrm{C}-6^{\prime}$ ); 134.9 ( $\mathrm{C}-5^{\prime}$ ); $170.4(\mathrm{C}=0)$; $171.0(\mathrm{C}=0)$; $171.5(\mathrm{C}=0)$.
(2S,3R)-2,3-Diacetamido-5-acetoxypentanoic Acid (17)
To a suspension of $15(0.10 \mathrm{~g}, 0.39 \mathrm{mmol})$ in $\mathrm{CCl}_{4}-\mathrm{MeCN}-\mathrm{H}_{2} \mathrm{O}(1: 1: 1,15 \mathrm{ml})$ were added $\mathrm{NaIO}_{4}(0.42 \mathrm{~g}, 2.10 \mathrm{mmol})$ and $\mathrm{RuCl}_{3} \cdot \mathrm{H}_{2} \mathrm{O}(1 \mathrm{mg})$. The reaction mixture was stirred at room temperature for 3 h , and the solid was removed by filtration. After evaporation of the sol-
vent under reduce pressure, the crude aldehyde $\mathbf{1 6}$ (9.59 ppm for $\mathrm{HCO}, 197.8 \mathrm{ppm}$ for $\mathrm{C}=0$ ) was used in the subsequent reaction without purification.

A solution of $\mathrm{NaClO}_{2}(162 \mathrm{mg}, 1.79 \mathrm{mmol}, 80 \mathrm{wt} . \%)$ and $\mathrm{NaH}_{2} \mathrm{PO}_{4} \cdot 2 \mathrm{H}_{2} \mathrm{O}$ ( 205 mg , 1.31 mmol ) in 1 ml of water was added dropwise to a solution of aldehyde $\mathbf{1 6}$ ( 50 mg , $0.194 \mathrm{mmol})$ in acetonitrile-tert-butyl alcohol-2-methylbut-2-ene (4:4:1, 4 ml ) at $0{ }^{\circ} \mathrm{C}$ over 5 min and the resulting solution was stirred at the same temperature for 1 h . Then the reaction mixture was poured into brine ( 1 ml ) and extracted with ethyl acetate ( $2 \times 10 \mathrm{ml}$ ). The combined organic layers were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated under reduced pressure. The residue was chromatographed on silica gel (dichloromethane-methanol, 3:1) affording $39 \mathrm{mg}(73 \%)$ of acid 17 as a colorless oil, $[\alpha]_{D}{ }^{25}+36$ (c $0.21, \mathrm{H}_{2} \mathrm{O}$ ). For $\mathrm{C}_{11} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}_{6}$ (274.3) calculated: $48.17 \% \mathrm{C}, 6.62 \% \mathrm{H}, 10.21 \% \mathrm{~N}$; found: $48.32 \% \mathrm{C}, 6.34 \% \mathrm{H}, 10.38 \% \mathrm{~N} .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{D}_{2} \mathrm{O}$ ): $1.71-1.80 \mathrm{~m}, 2 \mathrm{H}\left(\mathrm{H}-4^{\prime}\right) ; 1.98 \mathrm{~s}, 3 \mathrm{H}\left(\mathrm{CH}_{3}\right) ; 2.04 \mathrm{~s}, 3 \mathrm{H}\left(\mathrm{CH}_{3}\right) ; 2.07 \mathrm{~s}, 3 \mathrm{H}$ $\left(\mathrm{CH}_{3}\right) ; 4.10 \mathrm{~m}, 2 \mathrm{H}\left(\mathrm{H}-5^{\prime}\right) ; 4.27 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-2^{\prime}\right) ; 4.34 \mathrm{~m}, 1 \mathrm{H}\left(\mathrm{H}-3^{\prime}\right) .{ }^{13} \mathrm{C} N \mathrm{NR}\left(400 \mathrm{MHz}, \mathrm{D}_{2} \mathrm{O}\right)$ : $23.1\left(\mathrm{CH}_{3}\right) ; 24.7\left(2 \times \mathrm{CH}_{3}\right) ; 32.6\left(\mathrm{C}-4^{\prime}\right) ; 51.3\left(\mathrm{C}-3^{\prime}\right) ; 57.1\left(\mathrm{C}-2^{\prime}\right) ; 60.9\left(\mathrm{C}-5^{\prime}\right) ; 169.3(\mathrm{C}=0)$; 170.9 ( $\mathrm{C}=0$; $171.6(2 \times \mathrm{C}=0)$.

This work was supported by the Grant Agency (projects $1 / 0446 / 03$ and $1 / 2472 / 05$ ) of the Ministry of Education, Slovak Republic. NMR experiments were supported by the Ministry of Education, Slovak Rebublic (project 200280203/2003).

## REFERENCES

1. a) Williams R. M.: Synthesis of Optically Active $\alpha$-Amino Acids. Pergamon, Oxford 1989; b) Calmes M., Daunis J.: Amino Acids 1999, 16, 215.
2. a) Nomoto S., Teshima T., Wakamiya T., Shiba T.: Tetrahedron 1978, 34, 921; b) Yoshioka H., Aoki T., Goko H., Nakatsu K., Noda T., Sakakibara H., Take T., Nagata A., Abe J.: Tetrahedron Lett. 1971, 12, 2043.
3. a) Bycroft B. W., Cameron D., Johnson A. W.: J. Chem. Soc. C 1971, 3040; b) Wakamiya T., Katsutoshi M., Tatsuzo U., Teshima T., Shiba T.: Bull. Chem. Soc. Jpn. 1978, 51, 850; c) Shiba T., Ukita T., Mizuno K., Teshima T., Wakamiya T.: Tetrahedron Lett. 1977, 18, 2681; d) Teshima T., Konishi K., Shiba T.: Bull. Chem. Soc. Jpn. 1980, 53, 508; e) De Mong D. E., Willams R. M.: Tetrahedron Lett. 2001, 42, 3529; f) Jackson M. D., Gould S. J., Zabriskie T. M.: J. Org. Chem. 2002, 67, 2934; g) De Mong D. E., Williams R. M.: J. Am. Chem. Soc. 2003, 125, 8561.
4. Martinková M., Gonda J., Džoganová M.: Chem. Pap. 2004, 58, 292.
5. a) Martinková M., Gonda J.: Tetrahedron Lett. 1997, 38, 875; b) Gonda J., Martinková M., Imrich J.: Tetrahedron 2002, 58, 1611.
6. Koch T., Buchardt O.: Synthesis 1993, 1065.
7. a) Gmeiner P., Junge D., Kärtner A.: J. Org. Chem. 1994, 59, 6766; b) Sendzik M., Guarnieri W., Hoppe D.: Synthesis 1998, 1287.
8. Frigerio M., Santagostino M., Sputore S., Palmisano G.: J. Org. Chem. 1995, 60, 7272.
9. Kniežo L., Bernát J., Martinková M.: Chem. Pap. 1994, 48, 103.
10. Carlsen P. H. J., Katsuki T., Martin V. S., Sharpless K. B.: J. Org. Chem. 1981, 46, 3936.
11. Reginato G., Mordini A., Valacchi M., Grandini E.: J. Org. Chem. 1999, 64, 9211.
12. Pattenden G., Reynolds S. J.: J. Chem. Soc., Perkin Trans. 1 1994, 379.
13. Kang S. H., Hwang Y. S., Youn J.-H.: Tetrahedron Lett. 2001, 42, 7599.
14. a) Kessler P., Servent D., Hirth Ch.: Tetrahedron Lett. 1994, 35, 7237; b) Ksander G. M., Yuan R. J. A., Ghai R. D., Trapani A., McMartin C.: J. Med. Chem. 1997, 40, 495; c) Moon S.-H., Lee S.: Synth. Commun. 1998, 28, 3919.
15. Collier P. N., Campbell A. D., Patel I., Raynham T. M., Taylor R. J. K.: J. Org. Chem. 2002, 67, 1802.

[^0]:    Collect. Czech. Chem. Commun. 2006, Vol. 71, No. 8, pp. 1199-1210
    © 2006 Institute of Organic Chemistry and Biochemistry

[^1]:    Scheme 2

